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# Spectroscopic Studies of a Sensory Rhodopsin I Homologue from the Archaeon Haloarcula vallismortis<sup>†</sup>

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ABSTRACT: Sensory rhodopsin I (SRI) functions as a dual receptor regulating both negative and positive phototaxis. It transmits light signals through changes in protein-protein interactions with its transducer protein, HtrI. The phototaxis function of Halobacterium salinarum SRI (HsSRI) has been well characterized using genetic and molecular techniques, whereas that of Salinibacter ruber SRI (SrSRI) has not. SrSRI has the advantage of high protein stability compared with HsSRI and, therefore, provided new information about structural changes and Cl<sup>-</sup> binding of SRI. However, nothing is known about the functional role of SrSRI in phototaxis behavior. In this study, we expressed a SRI homologue from the archaeon *Haloarcula vallismortis* (HvSRI) as a recombinant protein which uses all-trans-retinal as a chromophore. Functionally important residues of HsSRI are completely conserved in HvSRI (unlike in SrSRI), and HvSRI is extremely stable in buffers without Cl<sup>-</sup>. Taking advantage of the high stability, we characterized the photochemical properties of HvSRI under acidic and basic conditions and observed the effects of  $Cl^-$  on the protein under both conditions. Fourier transform infrared results revealed that the structural changes in HvSRI were quite similar to those in HsSRI and SrSRI. Thus, HvSRI can become a useful protein model for improving our understanding of the molecular mechanism of the dual photosensing by SRI.

Numerous microorganisms have rhodopsins in their cell membranes, serving as photoreceptors. Bacteriorhodopsin (BR)<sup>1</sup> is one of the most famous membrane proteins and functions as a light-driven proton pump (1). The high expression level of BR and the formation of the stable membrane patches helped in the determination of its atomic structure (1), which has been used as a guide to analyze the properties and mechanisms of other BR-like proteins and other types of rhodopsins as well. Sensory rhodopsin I (SRI) and sensory rhodopsin II (SRII) form 2:2 complexes with their cognate transducer proteins, halobacterial transducer for SRI (HtrI) and halobacterial transducer for SRII (HtrII), respectively (2, 3), and their complexes work as sensors to transfer signals to the cytoplasm. SRII is specialized as a negative phototaxis receptor (4-6), while SRI functions as a receptor regulating both negative and positive phototaxis in

Halobacterium salinarum (7-9). Because of its high expression level in Escherichia coli and stability (10-12), crystallization of Natronomonas pharaonis SRII (NpSRII, also called ppR) and its cocrystallization with the truncated HtrII were achieved (2, 13, 14). Also, a number of experiments using electrophysiology (15) and EPR (16) have been performed on NpSRII, which would not have been possible using the homologous protein from H. salinarum.

The dark state of HsSRI and its long-lived M-intermediate are responsible for positive and negative phototaxis, respectively, and have absorption maxima at 587 and 373 nm, respectively (7). HsSRI is unstable in dilute salt solutions (17); therefore, little is known about the molecular mechanisms of interactions between SRI and HtrI, about structural changes, or about the signal relay mechanism. Recently, we characterized a new SRI protein from the eubacterium Salinibacter ruber (SrSRI) (17). SrSRI is much more stable under various conditions, and therefore, it provided new information about protein structural changes (18) and Cl<sup>-</sup> binding (19). However, no information is available about the functional role of SrSRI in phototaxis, although S. ruber cells show negative and positive phototaxis.

In this study, we cloned a gene encoding SRI-like protein from Haloarcula vallismortis (HvSRI), which has been discovered as csR3 and sequenced by Kitajima et al. in 1996 (20), expressed it in E. coli, and succeeded in purifying the highly stable protein. The sequence of HvSRI contains all amino acid residues (N165, H166, D201, and R215) identified as necessary for proper

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\*To whom correspondence should be addressed. H.K.: telephone and fax, 81 52 735 5207; e-mail, kandori@nitech.ac.jp. Y.S.: telephone, 81 52 789 2993; fax, 81 52 789 3001; e-mail, z47867a@nucc.nagoya-u.ac.jp. Abbreviations: BR, bacteriorhodopsin; DDM, *n*-dodecyl β-D-malto-

side; HPLC, high-performance liquid chromatography; HvSRI, SRIlike protein from Haloarcula vallismortis; HvSRIK, K-intermediate of HvSRI; HvSRI<sub>M</sub>, M-intermediate of HvSRI; HvSRI<sub>P</sub>, P-intermediate of HvSRI; PG, L-α-phosphatidylglycerol; SRI, sensory rhodopsin I.

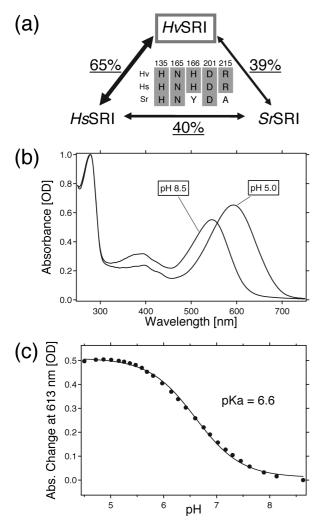


FIGURE 1: (a) Sequence similarities among the three SRI species. Percentages represent the levels of identity between SRI sequences estimated with ClustalW. Five residues important for the phototaxis function are shown in the middle. Residue numbers shown are for HvSRI (GenBank entry D83748). Other accession numbers: HsSRI, H. salimarum (L05603); SrSRI, S. ruber (3852586). (b) Absorption spectra of HvSRI. The PG-reconstituted sample was resuspended in citric acid buffer (pH 5.0) with 1 M NaCl or in borate buffer (pH 8.5) with 1 M NaCl. The temperature was kept at 25 °C. (c) pH titration curve of the counterion, Asp76, in HvSRI. The titration curve was analyzed using the Henderson—Hasselbalch equation with a single  $pK_a$  value.

phototaxis in HsSRI (9, 21-23) as well as a histidine residue (H135) important for chloride ion binding (19) (Figure 1a). The degree of identity of HvSRI with HsSRI is 65%, which is much higher than that between HsSRI and SrSRI (40%) (17). Even with this high degree of homology, the difference in the protein stability with HsSRI is substantial. Utilizing the high stability, we performed the first characterization of HvSRI under acidic and basic conditions.

### MATERIALS AND METHODS

Plasmids and Strains. Ha. vallismortis was grown aerobically at 37 °C and pH 7.0 in medium with the following composition: 250 g/L NaCl, 20 g/L MgSO<sub>4</sub>·7H<sub>2</sub>O, 2 g/L KCl, 0.2 g/L CaCl<sub>2</sub>, 3.0 g/L Na<sub>3</sub>(citrate), and 10 g/L Oxoid pepton (Oxoid Ltd., Hampshire, U.K.). E. coli DH5α and BL21(DE3) were used as hosts for DNA manipulation and for protein expression, respectively. The SRI gene was amplified using PCR from the genomic DNA (24). Primers were designed by

appending an *Nde*I site to the start codon and an *Xho*I site in apposition of the stop codon. The *Nde*I—*Xho*I fragment was digested and ligated to the *Nde*I and *Xho*I sites of the pET21c(+) vector (Novagen, Madison, WI). Consequently, the plasmid encodes six histidines at the C-terminus. This cloning strategy resulted in the following N- and C-terminal peptide sequence: <sup>1</sup>MDAV-----TPAD<sup>236</sup>LEHHHHHHH. The constructed plasmid was analyzed using an automated sequencer to confirm the expected nucleotide sequences.

Protein Expression, Purification, and Reconstitution into PG Liposomes. The HvSRI protein sample was prepared as described previously (17). Briefly, the protein was expressed in BL21(DE3) cells, solubilized with 1.0% n-dodecyl β-D-maltoside (DDM), and then purified using a Ni<sup>2+</sup>-NTA column (QIAGEN, Valencia, CA) (25). The purified samples were then reconstituted into PG liposomes (1:30 HvSRI:PG molar ratio) via removal of the detergent with Bio-Beads (SM2, Bio-Rad, Hercules, CA) (25). To monitor the effects of salt on the stability of HvSRI, the PG-reconstituted sample was resuspended in a buffer containing 50 mM Tris-HCl (pH 7.0). Samples were kept at 20 °C, and the absorption spectra were recorded at different time intervals.

pH Titration and High-Performance Liquid Chromatography Analysis. UV-vis spectra were recorded at room temperature using a UV2400PC spectrophotometer with an ISR240A integrating sphere (Shimadzu, Kyoto, Japan). The PG-reconstituted samples were washed and resuspended in a six-component buffer (citric acid, MES, HEPES, MOPS, CHES, and CAPS, whose concentrations were 10 mM each) with 1 M NaCl. The pH of each sample was measured using an F-55 pHmeter with a 9669-C10 glass electrode (HORIBA, Kyoto, Japan). High-performance liquid chromatography (HPLC) analysis was performed as described previously (19). The PG-reconstituted sample was analyzed in buffer containing 1 M NaCl and 10 mM citric acid (pH 5.0) or 10 mM borate (pH 8.5). The molar compositions of retinal isomers were calculated from areas of the peaks monitored at 360 nm. The assignment of each peak was performed by comparing it with the HPLC pattern from retinal oximes of authentic all-trans- and 13-cis-retinals (26).

Low-Temperature UV-Vis Spectroscopy. Low-temperature UV-vis spectra were recorded using a V-550DS spectrophotometer (JASCO, Tokyo, Japan) with a cryostat (Optistat DN, Oxford, U.K.) and a temperature controller (ITC502) with liquid nitrogen as the coolant, as described previously (17). Each PG-reconstituted sample with 2 mM citric acid (pH 5.0) or 2 mM borate (pH 8.5) was placed on a BaF2 window and then was dried to a film under pressure reduced with an aspirator. The films were then hydrated with water. To calculate the HvSRI<sub>M</sub> minus HvSRI and the HvSRI<sub>P</sub> minus HvSRI<sub>M</sub> difference spectra, measurements were performed at 200-260 K as follows. The HvSRI film was illuminated with > 500 nm light (Y-52) for 2 min, which converts HvSRI to HvSRI<sub>M</sub>, and then was illuminated through a UV filter (~400 nm light, UG-5, Melles Griot) for 1 min to convert HvSRI<sub>M</sub> to HvSRI<sub>P</sub>. The difference spectra were calculated from the spectra before and after each illumination.

Time-Resolved Flash Spectroscopy. For the flash photolysis experiment on the microsecond time scale (decay of the K-intermediate), the apparatus and the procedure were essentially as described previously (19). The purified sample was resuspended in a buffer containing 10 mM citric acid (pH 5.0) or 10 mM borate (pH 8.5) with 1 M NaCl and 0.1% DDM and

was placed in a quartz cell, and then the absorbance was adjusted to  $\sim \! 0.5$  OD unit (optical pass length of 2 mm) at the excitation wavelength. The temperature was kept at 25 °C during the measurement. For measurements of the M-intermediate, the conventional flash photolysis apparatus and procedure were essentially as described previously (27). The purified sample was resuspended in a buffer containing 10 mM citric acid (pH 5.0) or 10 mM borate (pH 8.5) with 1 M NaCl and 0.1% DDM. Although a phototransient signal could be acquired from a single flash, several kinetic traces were averaged to improve the signal-to-noise (S/N) ratio.

Low-Temperature Fourier Transform Infrared (FTIR) Spectroscopy. Low-temperature FTIR spectroscopy was performed as described previously (18). After hydration with H<sub>2</sub>O or D<sub>2</sub>O, the sample film was placed in a cell mounted in an Oxford Optistat-DN cryostat placed in an FTS-40 spectrometer (Bio-Rad) for HvSRI<sub>K</sub> or an FTS-7000 spectrometer (DIGILAB) for HvSRI<sub>M</sub>. Spectra were constructed from 128 interferograms with a spectral resolution of 2 cm<sup>-1</sup>. Difference spectra were calculated from spectra before and after illumination. Illumination with 500 nm light at 77 K for 2 min converted HvSRI to HvSRI<sub>K</sub>, and a subsequent illumination with > 610 nm light (R63) reverted the HvSRIK back to HvSRI. Thirty-two difference spectra obtained in this way were averaged for each HvSRIK minus HvSRI spectrum. Illumination with >500 nm light (Y52) at 260 K for 2 min converted HvSRI to HvSRI<sub>M</sub>. Six difference spectra obtained in this way were averaged for each HvSRI<sub>M</sub> minus HvSRI spectrum.

## **RESULTS**

Protein Expression, Purification, and UV-Vis Spectroscopy. To determine whether the HvSRI gene encodes a functional protein, we cloned and expressed HvSRI in E. coli BL21-(DE3) cells. Induction with 0.5 mM isopropyl 1-thio- $\beta$ -D-galactopyranoside and the addition of retinal (5  $\mu$ M) changed the color of the transformed cells. Purified HvSRI was reconstituted into PG liposomes, and absorption spectra at pH 5.0 and 8.5 were recorded, as shown in Figure 1b. The absorption maximum of HvSRI is located at 593 nm at pH 5.0 and at 545 nm at pH 8.5. The spectral shift must be due to the protonation of Asp76 of HvSRI, because in all retinal proteins, except for halorhodopsin (HR), the protonated Schiff base is stabilized by an aspartic acid as a counterion (1, 28-30). These maximum wavelengths are similar to those of HsSRI (587 and 552 nm, respectively) (31) and SrSRI (620 and 558 nm, respectively) (17) (Table 1). Alignment of the amino acid sequences with ClustalW reveals that HvSRI is 65 and 39% identical to HsSRI and SrSRI, respectively (Figure 1a). Thus, we were able to express HvSRI as a photoactive membrane-embedded protein.

Retinal Configuration and  $pK_a$  of the Counterion. To investigate the properties of HvSRI, HPLC and pH titration experiments were performed. While light-driven ion pumps (e.g., BR and HR) possess both all-trans-retinal and 13-cis-retinal in the dark (32, 33), the photosensors HsSRI, HsSRII, and NpSRII have only all-trans-retinal (26, 34). The isomeric state of retinal in HvSRI was predominantly all-trans under both acidic and basic conditions (~95.9%), and the retinal configuration was not altered by illumination (480 nm light for 1 min with a Xe lamp), indicating the absence of light or dark adaptation. Thus, the retinal composition of HvSRI is markedly sensory rhodopsin-like (Table 1).

Table 1: Comparison of Characteristics of SRIs

	$\lambda_{max}$ (acid/base) (nm)	retinal composition	$pK_a$	stability	Cl <sup>-</sup> effects
HvSRI HsSRI	593/545 587/552	all- <i>trans</i> all- <i>trans</i>	6.6 7.2	high low	yes not determined
SrSRI	$\sim 620/558$	all- <i>trans</i>	4.3	high	yes

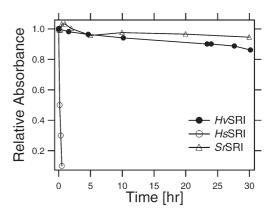


FIGURE 2: Time-dependent absorbance changes of SRI proteins in the absence of NaCl. The samples were reconstituted into PG liposomes and were suspended in a buffer containing 50 mM Tris-HCl (pH 7.0), and the temperature was maintained at 20 °C. The plots for *Hs*SRI and *Sr*SRI were reproduced from previous studies (17).

To determine the  $pK_a$  value of the counterion Asp76 of HvSRI, we performed pH titration experiments. The titration curve was analyzed using the Henderson-Hasselbalch equation with a single  $pK_a$  value, and the value in the unphotolyzed state was estimated to be 6.6 (Figure 1c). This is much closer to that of HsSRI (~7.2) (35) than to that of SrSRI (4.3) (17) (Table 1), suggesting a structural similarity around the chromophore between HvSRI and HsSRI, but not SrSRI. Because the counterion  $pK_a$  values of HvSRI and HsSRI are at neutral pH where these bacteria live, both acidic and basic forms may be considered important for the phototaxis function. Therefore, we characterized HvSRI under both acidic and basic conditions.

Protein Stability. Because of the instability of HsSRI in dilute salt solutions (17), little is known about the molecular mechanism for positive phototaxis. The SrSRI protein we characterized previously is highly stable under various conditions compared with HsSRI (17). Therefore, SrSRI is useful for characterization of the structural changes of the protein (18) and Cl<sup>-</sup> binding (19). On the other hand, it has several differences from HsSRI, i.e., lack of some functionally important residues (Figure 1a), a lower absorption maximum, a faster overall cycle, and a lower  $pK_a$  value of the Schiff base counterion (Table 1). Nevertheless, no structure of any SRI has been determined. Figure 2 shows the stability of HvSRI compared with HsSRI and SrSRI under salt-free conditions, as judged by the light absorbance of the proteins. Fast degradation of HsSRI is obvious, whereas surprisingly, HvSRI exhibited almost the same stability as SrSRI. In addition, PG-reconstituted HvSRI is stable over a wide pH (pH 4-9.5) and temperature ( $\sim$ 60 °C) range even in the absence of NaCl and is also expressed in E. coli at higher levels ( $\sim$ 8 mg/L of culture) than SrSRI ( $\sim$ 5 mg/L of culture).

Photocycle Kinetics. Utilizing the high stability and the high expression level, we characterized the photocycle of HvSRI by using low-temperature UV—vis spectroscopy (Figure 3) and flash

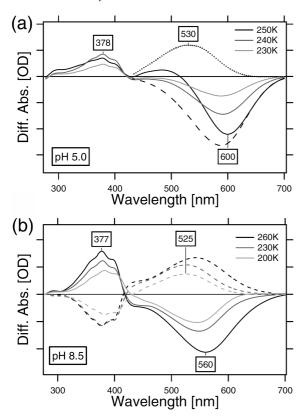


FIGURE 3: Low-temperature UV-vis difference absorption spectra of light-activated HvSRI. The PG-reconstituted sample was suspended in citric acid (pH 5.0) or in borate (pH 8.5) without NaCl. (a) The difference absorption spectra of the pH 5.0 sample at three different temperatures were calculated from the spectra before and after irradiation with > 500 nm light. The dashed line is the simulated absorbance spectrum of the initial state of HvSRI fitted on the slope of the high-wavelength side of the light minus dark difference spectrum at 250 K. The dotted line is the light minus dark difference spectrum minus the simulated initial spectrum, whose peak appears at 530 nm. (b) The difference absorption spectra of the pH 8.5 sample at three different temperatures were calculated from the spectra before and after irradiation with > 500 nm light (—) and from the spectra after the > 500 nm irradiation and after the subsequent UV illumination (---). Each division of the y-axis of panels a and b corresponds to 0.05 and 0.1 absorbance unit, respectively.

photolysis (Figure 4) experiments. The estimated kinetic parameters are listed in Table 2. For the flash photolysis experiments, a DDM-solubilized sample was used to increase the S/N ratio without inactivation. Low-temperature UV-vis difference spectroscopy reveals that the acidic form of HvSRI has a positive peak at ~500 nm upon illumination at 250 K, which seems to equilibrate with the M-intermediate absorbing at 375 nm (Figure 3a). The light minus dark difference spectrum minus the simulated spectrum of the initial dark state shows the peak at 530 nm; therefore, we denote this intermediate as  $HvSRI_{530}$ . As the temperature decreases from 250 to 240 K, the HvSRI<sub>M</sub> spectral amplitude increases while that of HvSRI<sub>530</sub> decreases, which suggests that, although HvSRI<sub>M</sub> and HvSRI<sub>530</sub> are in equilibrium, the transition from HvSRI<sub>M</sub> to HvSRI<sub>530</sub> seems to be thermally more favorable. Therefore, HvSRI<sub>530</sub> is tentatively assigned as the N-intermediate of HvSRI. The basic form of HvSRI has an M-intermediate absorbing at 377 nm as shown in Figure 3b. Since the M-intermediate of basic HvSRI is very stable, we tested whether HvSRI<sub>M</sub> forms a P510-like intermediate of HsSRI which is important for negative phototaxis (7). UV illumination of the trapped M-intermediate converted it to a

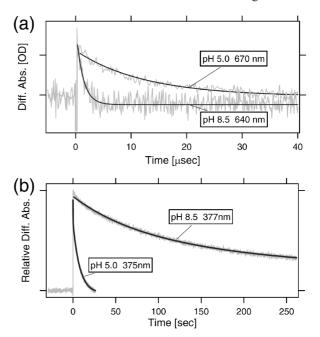


FIGURE 4: Flash-induced kinetic data of HvSRI dissolved in citric acid or in borate containing 1 M NaCl and 0.1% DDM at pH 5.0 or 8.5. (a) Absorbance changes at 670 or 640 nm represent the K-intermediate decay in the acidic or basic form of HvSRI, respectively. The monitoring wavelengths were chosen from the flashinduced difference absorption spectra of HvSRI at each pH over a spectral range from 480 to 725 nm on the nanosecond to microsecond time scale (data not shown). Both sets of data fit well to a singleexponential decay equation, and the half-times were estimated to be 12.7  $\mu$ s for pH 5.0 and 1.1  $\mu$ s for pH 8.5. The division of the y-axis corresponds to 0.01 absorbance unit. (b) Monitoring the absorbance changes at 375 nm and pH 5.0 or 377 nm and pH 8.5 gave the rates of decay of the M-intermediate. Since the maximal amplitude of the M-intermediate of the acidic form was smaller by a third compared to that of the basic form, the data for the acidic form were renormalized for visual comparison of the decay rates. For all the measurements, the temperature was maintained at 25 °C.

Table 2: Half-Lifetime Values for the K- and M-Intermediates half-lifetime value intermediate pН K 5.0  $12.7 \, \mu s$ 8.5  $1.1 \,\mu s$ M 5.0 6.2 s (65%) 1.2 s (10%) 54.7 ms (25%) 8.5 90.5 s (88%) 6.5 s (12%)

red-shifted product with a difference absorbance maximum at 525 nm (Figure 3b). A similar experiment was conducted with the acidic form of *Hv*SRI, and the blue shift was observed as well, although the data were obtained with a low S/N ratio (data not shown).

Using laser flash photolysis, we measured the transient absorption spectra at various time points after excitation on the microsecond time scale, and the wavelengths of maximum absorbance of the K-intermediates were found to be 670 nm at pH 5.0 and 640 nm at pH 8.5 (data not shown). Figure 4a shows the time traces of the absorption changes at 670 nm and pH 5.0 and at 640 nm and pH 8.5. Both curves fit well to a single-exponential decay equation, and the half-times were estimated to be 12.7  $\mu$ s at pH 5.0 and 1.1  $\mu$ s at pH 8.5, indicating that the decay rate of the K-intermediate of acidic  $H\nu$ SRI is almost the same as that of  $H\nu$ SRI at pH 6.0 (36) and that of basic  $H\nu$ SRI is 21.8-fold faster than that of SrSRI (19). The decay curve of the acidic form of  $H\nu$ SRI<sub>M</sub> at 375 nm fits well to a triple-exponential decay equation with half-time values of 6.2 s (65%), 1.2 s (10%), and

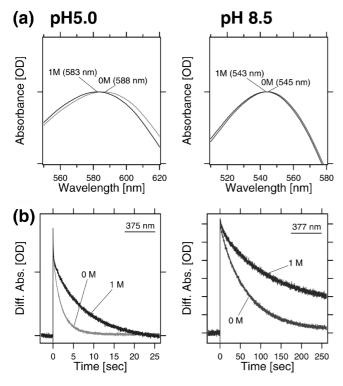


FIGURE 5: (a) Absorption spectra of HvSRI at pH 5.0 (left) or 8.5 (right). The samples was suspended in 0.1% DDM buffer with 1 M NaCl or 0 M NaCl (333 mM Na<sub>2</sub>SO<sub>4</sub>). The data are shown over a spectral range from 550 to 620 nm for pH 5.0 (from 510 to 580 nm for pH 8.5). (b) Strobe flash-induced kinetic data of HvSRI for pH 5.0 (left) or 8.5 (right) on the second time scale. The samples were suspended in 0.1% DDM buffer with 1 M NaCl or 0 M NaCl (333 mM Na<sub>2</sub>SO<sub>4</sub>). The curves were fitted by a double-exponential decay equation. The temperature was kept at 25 °C. Each division of the y-axis of panels a and b corresponds to 0.2 and 0.01 absorbance unit, respectively.

54.7 ms (25%) (Figure 4b and Table 2), while the decay curve of the basic form of  $HvSRI_M$  fits well to a double-exponential decay equation (Figure 4b). The estimated half-time values were 90.5 s (88%) and 6.5 s (12%) (Table 2).  $SrSRI_M$  has a 115.5 ms decay half-time under similar conditions, as in the basic form, which means that the M-intermediate decay of HvSRI in the basic form is approximately 800 times slower than that of SrSRI. According to Losi et al (36), DDM-dissolved HsSRI at pH 6.0, where its counterion is deprotonated, had lifetime value of 12 s for M-intermediate decay and the undeprotonatable mutant (D76N) had a value of 28 s. Comparing these values with the results for HvSRI described above, we could say that photocycle kinetics of HvSRI is fairly close to that of HsSRI rather than SrSRI.

Effects of Chloride on HvSRI. Figure 5 shows effects of Cl<sup>-</sup> on the absorption maxima (a) and photocycle kinetics (b) under acidic and basic conditions. The absorption maximum of HvSRI is shifted from 588 to 583 nm at pH 5.0 (-5 nm) and from 545 to 543 nm at pH 8.5 (-2 nm). As shown in Figure 5b, the M-intermediate decay rates increased by 3.4-fold at pH 5.0 and 1.7-fold at pH 8.5 in the absence of Cl<sup>-</sup> compared to those in a 1 M NaCl solution. In the case of SrSRI having a deprotonated counterion (the alkaline form), the  $\lambda_{\rm max}$  is shifted from 542 to 556 nm (+14 nm) in a Cl<sup>-</sup>-dependent manner (19). The M-intermediate decay rate is also increased 5-fold in the absence of Cl<sup>-</sup> compared to that in a 1 M NaCl solution (19). It should be noted that it is hard to observe the effects of Cl<sup>-</sup> on the photochemical properties of SrSRI under the acidic condition (pH < 4.3) because of its instability.

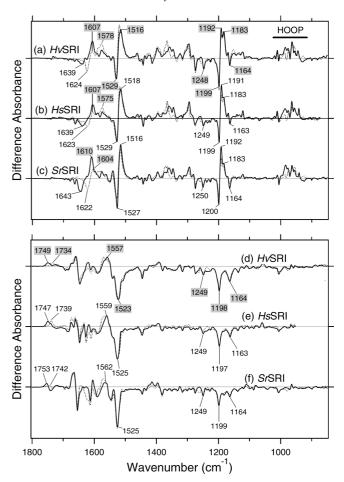


FIGURE 6: (a) Difference infrared spectra ( $HvSRI_K$  minus HvSRI) measured at 77 K and pH 8.5 in the 1800-850 cm<sup>-1</sup> region. Each sample was hydrated with  $H_2O$  (—) or  $D_2O$  (---). The spectra in panels a and b were multiplied by 4.2 for the sake of comparison. The spectra of HsSRI (b) were deleted at < 872 cm<sup>-1</sup>. The spectra in panels b and c were reproduced from ref 18 for the sake of comparison. Each division of the y-axis corresponds to 0.015 absorbance unit. (d) Difference infrared spectra ( $HvSRI_M$  minus HvSRI) measured at 260 K and pH 8.5 in the 1800-850 cm<sup>-1</sup> region. Each sample was hydrated with  $H_2O$  (—) or  $D_2O$  (---). The spectra in panel d were multiplied by 0.8 for the sake of comparison. The spectra of HsSRI (e) were deleted at < 951 cm<sup>-1</sup>. The spectra in panels e and f were reproduced from ref 18 for the sake of comparison. Each division of the y-axis corresponds to 0.012 absorbance unit.

FTIR Spectroscopy. To analyze the structural changes of HvSRI during the photocycle, FTIR spectroscopy was performed. The top panel in Figure 6 shows K minus initial state difference FTIR spectra of HvSRI in the 1800–850 cm<sup>-1</sup> region. The HsSRI<sub>K</sub> minus HsSRI and SrSRI<sub>K</sub> minus SrSRI spectra are reproduced from ref 18 for the sake of comparison. The bands at 1529(-)/1516(+) cm<sup>-1</sup> for HvSRI correspond to the ethylenic C=C stretching vibrations of the retinal chromophore. The lower-frequency shift corresponds to the spectral red shift upon formation of the K-intermediate. The bands at 1248(-), 1199(-)/1192(+), and 1183(+)/1164(-) cm<sup>-1</sup> for HvSRI correspond to the C–C stretching vibrations of the retinal chromophore. The 1639 and 1624 cm $^{-1}$  bands for HvSRI correspond to the C=N stretching vibrations in H<sub>2</sub>O and D<sub>2</sub>O, respectively. The difference in frequencies between H<sub>2</sub>O and D<sub>2</sub>O has been regarded as a marker of the strength of hydrogen bonding between the Schiff base and the counterion (37-39). Frequency differences are 15 cm<sup>-1</sup> in HvSRI, 16 cm<sup>-1</sup> in HsSRI (40), and 21 cm<sup>-1</sup> in SrSRI (18), all of which are close to the value for

NpSRII, 17 cm<sup>-1</sup> (25). In the K-intermediate, this C=N stretching vibration frequency shifts to 1607 cm<sup>-1</sup> for HvSRI<sub>K</sub>, to 1607 cm<sup>-1</sup> for  $HsSRI_K$ , and to 1610 cm<sup>-1</sup> for  $SrSRI_K$  in  $H_2O$ . The downshift by D<sub>2</sub>O was observed only with SrSRI<sub>K</sub> (from 1610 to 1604 cm<sup>-1</sup>). However, only  $HvSRI_K$  and  $HsSRI_K$  in  $D_2O$ have positive bands at 1578 and 1575 cm<sup>-1</sup>, respectively, which could be candidates for the C=N stretch vibrations in  $D_2O$ .

The bottom panel of Figure 6 shows the M-intermediate minus initial state difference FTIR spectra of HvSRI in the 1800–850 cm<sup>-1</sup> region. The HsSRI<sub>M</sub> minus HsSRI and the SrSRI<sub>M</sub> minus SrSRI spectra are reproduced from ref 18 for the sake of comparison. The bands at 1557(+)/1523(-) cm<sup>-1</sup> for HvSRIcorrespond to the ethylenic C=C stretching vibrations of the retinal chromophore, and the amide II vibrations also possibly appear here. The negative bands at 1249, 1198, and 1164 cm<sup>-1</sup> for HvSRI are attributed to the C-C stretching vibrations of the retinal chromophore, where the intensities of the corresponding positive peaks seen in the K-intermediate are much reduced due to the loss of charge at the Schiff base upon deprotonation. When the Schiff base counterion is deprotonated, formation of the M-intermediate accompanies the transfer of a proton from the Schiff base to the counterion. In HvSRI, positive bands at 1749 cm $^{-1}$  in H<sub>2</sub>O and at 1734 cm $^{-1}$  in D<sub>2</sub>O were observed. Therefore, it is reasonable to assign these bands as the C=O stretching vibrations of Asp76. Thus, the overall signatures of the three SRI spectra upon formation of K- and M-intermediates look very similar to each other, indicating a common structure and structural changes among SRIs upon formation of the K- and M-intermediates.

### DISCUSSION

High Stability of HvSRI. Since the initial discovery of SRI in 1982 (41, 42), the signal relay mechanism from SRI to its cognate Htr transducer has become a focus of interest, in part because of its importance to the general understanding of communication between integral membrane proteins, about which little is known. Important advances in understanding the molecular mechanism of negative phototaxis regulated by the SRII—HtrII complex have been made in recent years (2, 6, 15, 16). In contrast, little is known about the molecular mechanism of positive phototaxis. In fact, high-resolution crystal structures of BR, HR, SRII, and the SRII-HtrII complex were reported from 1997 through 2006 (2, 13, 14, 43, 44), whereas the structure of SRI has not yet been determined. In 2008, we found and characterized a highly stable SRI homologous protein, SrSRI. However, SrSRI differs in several ways from HsSRI; e.g., SrSRI does not have the His residue on the F-helix conserved among the archaeal SRIs (Figure 1a), which may correlate with the absence of the acidic form of SrSRI, while HsSRI exists as the acidic form at neutral pH. Further, the photoreaction kinetics of free SrSRI are pH-independent, while those of the SrSRI-SrHtrI complex are pH-dependent, completely the opposite of HsSRI (27). In this study, we introduced a third SRI species, HvSRI, that has been successfully expressed and purified with good yield from E. coli cells as a recombinant protein with high stability (Figure 2). What is the difference that determines the stability of HvSRI in the absence of salt? In 1993, Oesterhelt and co-workers reported that truncation of the C-terminal region of HsSRI results in high levels of protein expression (45). Although the C-terminal region of HvSRI may be involved in protein stability, the amino acid sequence is quite different from those of HsSRI and SrSRI. HvSRI and SrSRI do not have a common component that is missing in HsSRI; thus, HvSRI must have a unique strategy for maintaining the integrity of its seven  $\alpha$ -helices. Since HvSRI and HsSRI are 65% identical, differences in their sequences are limited. Further study will be required to understand the high stability of HvSRI.

Comparison of Photochemical Properties of HvSRI with Those of HsSRI and SrSRI. We demonstrate here that the absorption maximum and p $K_a$  value of the counterion of HvSRIare quite similar to those of HsSRI, but not those of SrSRI (Figure 1 and Table 1). The photocycle kinetics of HvSRI are also more similar to those of HsSRI than of SrSRI. In HsSRI, it was reported that a histidine residue, His166, plays an important role in the pathways of proton transfer both to and from the Schiff base (22). The histidine residue is conserved in HvSRI but not in SrSRI (Figure 1a); therefore, this residue is a strong candidate for the determinant of the differences. In fact, the M-intermediate was observed even in the acidic form of HvSRI (Figure 3) where the counterion Asp76 is protonated (Figure 1). It could be explained by the transfer of a proton from the Schiff base of HvSRI<sub>593</sub> to the His166 homologue upon formation of the M-intermediate, like in HsSRI. In addition, we observed here the effects of Cl<sup>-</sup> on the absorption maximum and photocycling rate of HvSRI both under acidic and alkaline conditions (Figure 5). The acceleration of M-intermediate decay and the shift of the absorption maximum without Cl<sup>-</sup> are also observed in SrSRI, although HvSRI shows the Cl<sup>-</sup>-dependent blue shift that is different from that of SrSRI [14 nm red shift (19)]. In the case of SRII from N. pharaonis, no effect of Cl<sup>-</sup> on absorption maximum (46) or photocycle kinetics (47) was reported previously, although the spectral shift was observed under the acidic conditions (pH  $\leq$  3) in a chloride-dependent manner (46). These data strongly suggest that binding of Cl<sup>-</sup> to sensory rhodopsin I is widely conserved among members of the sensory rhodopsin I protein family, unlike in SRII, and assumed to be an important property for the functional differentiation between SRI and SRII. In fact, the putative Cl<sup>-</sup> binding site, His135, is also conserved in HsSRI and SrSRI (Figure 1a). The binding seems to be independent of the protonation state of Asp76 (Figure 5).

Structural Changes Revealed by FTIR Spectroscopy. We compared the hydrogen-out-of-plane (HOOP) vibration region  $(1050-850 \text{ cm}^{-1})$  of  $HvSRI_K$  minus the HvSRI difference FTIR spectra with those of HsSRI, SrSRI, NpSRII, BR, and the BR-T mutant. We have reported that the triple mutant, BR-T, is able to form a signaling complex with HtrII and can mediate phototaxis responses like SRII (48); therefore, BR-T is regarded as an SRII-like protein. Several peaks appear upon formation of the K-intermediate of HvSRI as well as HsSRI, SrSRI (Figure 6a), SRII (25), and BR-T (49), suggesting that the structural changes in the retinal chromophore seem to be essential for the sensory rhodopsins. Thus, the low-temperature FTIR spectroscopy upon formation of K- and M-intermediates showed that the difference spectra of HvSRI are quite similar to those of HsSRI as well as those of SrSRI (Figure 6), in particular with the structural changes of the retinal chromophre (HOOP, C=C, and C-C vibrations). However, in the region above 1600 cm<sup>-1</sup>, all three spectra exhibited differences probably due to the large effects of amide bands derived from the different protein moieties among three SRIs, especially in M-intermediates.

In conclusion, we have succeeded in purifying a highly stable SRI-like protein from *Ha. vallismortis*, and it can be expressed well in E. coli as a recombinant protein. The functionally important residues of HsSRI are completely conserved in HvSRI

(unlike in SrSRI). Spectroscopic studies revealed that the photochemical properties and structural changes of HvSRI are more similar to those of HsSRI than to those of SrSRI. Utilizing the high stability, the characteristics of HvSRI under the acidic and basic conditions were investigated, and Cl<sup>-</sup> effects were observed under both acidic and basic conditions. Thus, HvSRI will become a useful protein for further elucidation of the molecular mechanism of the dual photosensing by SRI. In addition to the cloning and expression of HtrI for HvSRI, further studies are required for investigation of the SRI-HtrI system.

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